
Mole Problems Answers

key - cp - mixed mole problems - mixed mole problems - key 1. a) how many grams are there in 1.5×10^{25} molecules of CO_2 ? 1.110g 1 mol 44.0 g 6.0210molecules 1.510molecules CO_2 1mol 3 23 2 25!=!!!! b) what volume would the CO_2 occupy at stp? 5.610l or 560 l 1 mol 22.4 l 6.0210molecules 1.510molecules CO_2 1mol 2 23 2 25!=!!!! 2. a) a sample of NH_3 gas occupies 75.0 liters at stp ... **practice problems: solutions (answer key) - clarkchargers** - calculate the mole fraction, molarity and molality of NH_3 if it is in a solution composed of 30.6 g NH_3 in 81.3 g of H_2O . the density of the solution is 0.982 g/ml and the density of water is 1.00 **2 3 2 3 2 1 2h3o2 3 2 ? f.u. nacl = 3.55 mol nacl 6.02 h ...** - part 2: mole conversions work each of the following problems. show all work. 1. how many atoms are in 6.2 moles of aluminum? ? atoms al = 6.2 mol al ! 6.02!10²³ atoms al 1 mol al = 3.7!10²⁴ atoms al 2. convert 5.3×10^{25} molecules of CO_2 to moles. ? moles CO_2 = 5.3!10²⁵ molecules CO_2 ! 1 mole CO_2 6.02!10³moleculs CO_2 = 88 mol CO_2 3. **chemistry: a study of matter - weebly** - worksheet: mole/mole problems name _____ chemistry: a study of matter © 2004, gpb 8.7a key 7.2 moles no 2 answer each of the following questions using the equation ... **worksheet: more mole problems name** - chemistry: a study of matter © 2004, gpb 7.11b 7. convert 2.76 g se to moles se. ? mol se = 2.76 g se ! 1 mol se 79.0 g se = 0.0349 mol se 58.5 g nacl **mole review practice problems** - 1 mole/ # grams 1 mole = 6.02×10^{23} atoms & molecules conversion problems: for each problem you must use dimensional analysis & box your answer. 7. given 12 g of aluminum , how many moles do you have? = 2992 26.98 ... mole review practice problems author: amanda hargrove **molar mass worksheet answer key - river dell regional ...** - molar mass worksheet - answer key calculate the molar masses of the following chemicals: 1) Cl_2 71 g/mol 2) KOH 56.1 g/mol 3) BeCl_2 80 g/mol 4) FeCl_3 162.3 g/mol 5) BF_3 67.8 g/mol 6) CCl_2F_2 ... "grams/mole" is better, because any macroscopic amount of a substance is better expressed in grams than amu. **2 mol al 27.0 g al worksheet: mole/mass problems name k ...** - balance each equation before solving any 3 m problems. show all work. 1. in a reaction between the elements aluminum and chlorine, aluminum chloride is produced. $\text{Al}_2\text{Cl}_6 + \text{Cl}_2 \rightarrow \text{AlCl}_3$ 2 a. 2 moles of al will react with _____ mole(s) of Cl_2 to produce _____ mole(s) of AlCl_3 . b. how many grams of AlCl_3 will be produced if 2.50 moles of ... **mole review practice problems - whitney high school** - conversion problems: for each problem you must use dimensional analysis, sig figs, ... 14. which weighs more, a mole of copper or a mole of gold? (write down how much each one weighs) 15. a. how are the terms "molar mass" and "atomic mass" different from each other? ... mole review practice problems author: amanda hargrove

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